

Hydroxyl group-containing diene rubbers

Sub A1
Ins A2

The present invention provides rubber mixtures which contain a diene rubber with a concentration of primary hydroxyl groups of 0.1 to 2 wt.% and a glass transition temperature of -120 to -50°C and their mixtures with fillers, optionally other rubbers and rubber auxiliary substances and vulcanisates prepared therefrom. Rubber mixtures according to the invention are suitable for producing highly reinforced, abrasion-resistant moulded items, in particular for producing tires which have particularly high wet skidding resistance.

40 Ins A3

Double bond-containing solution rubbers, such as solution polybutadiene and solution styrene/butadiene rubbers, have advantages over the corresponding emulsion rubbers when producing low rolling resistance tire treads. The advantages are based, inter alia, on the ability to control the vinyl content, cis content and glass transition temperature and molecular branching associated therewith. Particular advantages in relation to abrasion, wet skidding resistance and rolling resistance of the tires result therefrom in a practical application. Thus, US-A- 5,227,425 describes the production of tire treads from a solution SBR rubber and silica. To further improve the properties, numerous methods for modifying the end groups have been developed, as is described e.g. in EP-A 334 042, with dimethylaminopropyl-acrylamide or, as described in EP-A 447,066, with silyl ethers. However, due to the high molecular weight of the rubber, the proportion by weight of end groups is small and can therefore have only a small effect on the interaction between filler and rubber molecule. The present invention is intended to provide solution diene rubbers such as solution polybutadiene and polyisoprene with a much higher concentration of effective groups for interacting with fillers.

A process for preparing hydroxyl group-containing solution polybutadiene rubbers is also described in DE-OS 2,653,144. These rubbers, however, have a much higher concentration of hydroxyl groups, associated with much higher glass transition temperatures.

"Express Mail" mailing label number 09/980615-011902

Date of Deposit November 2, 2001

I hereby certify that this paper or fee is being deposited with the United States Postal Service "Express Mail Post Office to Addressee" service under 37 CFR 1.10 on the date indicated above and is addressed to the Assistant Commissioner of Patents and Trademarks, Washington, D.C. 20231

Donna J. Yeatch

(Name of person mailing paper or fee)

Donna J. Yeatch

09/980615-011902

EP-A 464,478 describes a process for hydroxylating rubber, wherein, however, secondary hydroxyl groups are introduced which are far less effective than the primary hydroxyl groups in the present invention.

5 Hydroxyl group-containing emulsion and solution rubbers are also described in EP-805,452 A1, wherein the hydroxyl concentrations described here for solution rubbers lie within a much lower range (0.009 to 0.061 %), depending on the process used, and the glass transition temperatures are substantially higher ($>-40^{\circ}\text{C}$), depending on the styrene content, for the emulsion rubbers described.

10

Ins 14

It has now been found that rubber mixtures and rubber vulcanisates with surprisingly improved dynamic damping characteristics in the temperature range relevant to wet skidding resistance and the temperature range relevant to rolling resistance and also improved abrasion behaviour can be prepared from hydroxyl group-containing solution diene rubbers with a concentration of 0.1 to 2 wt.% of bonded primary hydroxyl groups and a glass transition temperature of -120 to -50°C . Further surprising advantages were obtained when the rubber mixture was prepared not in conventional compounders but by mixing a solution of hydroxyl group-containing rubber and oxidic or siliceous fillers in organic solvents and the solvent was then removed using steam, since then the filler is fully precipitated with the rubber and does not remain in the waste water, as when using an unmodified rubber.

15

20

Ins 15

The present invention therefore provides rubber mixtures containing one or more hydroxyl group-containing rubbers built up from diolefins, characterised in that the hydroxyl group-containing rubber(s) contain in the range from 0.1 to 2 wt.% of bonded primary hydroxyl groups and have a glass transition temperature between -120 and -50°C and fillers. The invention also provides use of said rubber mixtures for preparing rubber vulcanisates, in particular silica-filled tire treads with especially high abrasion resistance, especially high wet skidding resistance and a low rolling resistance.

25

30

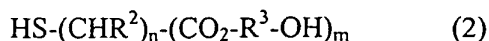
09980615-011802

Suitable diolefins are in particular 1,3-butadiene, isoprene, 1,3-pentadiene, 2,3-dimethylbutadiene, 1-phenyl-1,3-butadiene and/or 1,3-hexadiene. 1,3-butadiene and isoprene are particularly preferably used.

- 5 Rubbers according to the invention for use in rubber mixtures according to the invention can be prepared preferably by polymerisation using coordination catalysts in the presence of a solvent or by anionic solution polymerisation. Coordination catalysts in this connection are understood to be Ziegler-Natta catalysts, coordination catalysts and monometallic catalyst systems. Coordination catalysts are preferably
- 10 those based on Ni, Co, Ti or Nd. Catalysts for anionic solution polymerisation are based on alkali or alkaline earth metals such as e.g. n-butyllithium. In addition, known randomised control agents for the microstructure of the polymer may be used. These types of solution polymerisations are known and are described e.g. in I. Franta
- 15 Elastomers and Rubber Compounding Materials; Elsevier 1989, pages 113 - 131 and in Houben-Weyl, Methoden der Organischen Chemie, Thieme Verlag, Stuttgart, 1961, vol. XIV/1 pages 645 to 673 or in vol. E20 (1987), pages 114 to 134 and pages 134 to 153. The primary hydroxyl groups are introduced in a subsequent reaction of the final polymer. Methods for introducing primary hydroxyl groups are e.g. the addition of primary hydroxyl group-containing mercaptans, an addition reaction with
- 20 formaldehyde, reaction with carbon monoxide followed by hydrogenation, hydroboration followed by oxidative hydrolysis of the borane compound.

The hydroxyl groups are preferably introduced by the addition of hydroxylmercaptans of the general formula (1) and/or hydroxyl group-containing mercaptocarboxylic

25 esters of the general formula (2).



in which

R^1 represents a linear, branched or cyclic C_1 - C_{36} alkylene group, which may optionally be substituted with up to 6 further hydroxyl groups, or which may be interrupted by nitrogen, oxygen or sulfur atoms,

5 R^2 represents hydrogen or a C_1 - C_6 alkyl group and

R^3 represents a linear, branched or cyclic C_2 - C_{36} alkylene group, which may optionally be substituted with up to 6 further hydroxyl groups or may be interrupted by nitrogen, oxygen or sulfur atoms,

10

OH represents a primary hydroxyl group,

n is an integer from 1 to 5

15

m is an integer from 1 to 2.

C_1 - C_{36} alkylene groups are understood to be any linear, cyclic or branched alkylene groups with 1 to 36 carbon atoms which are known to a person skilled in the art, such as methylene, ethylene, n-propylene, i-propylene, n-butylene, i-butylene, t-butylene, n-pentylene, i-pentylene, neo-pentylene, n-hexylene, cyclohexylene, i-hexylene, heptylene, octylene, nonylene decylene, undecylene and dodecylene.

20

Preferred hydroxylmercaptans are mercaptoethanol, 1-mercapto-3-propanol, 1-mercapto-4-butanol, 3-mercapto-1,2-propanediol (thioglycerol), α -mercapto- ω -hydroxy-oligoethylene oxides such as e.g. α -mercapto- ω -hydroxyoctaethylene glycol or the corresponding ethylene oxide/propylene oxide mixed polyether. Mercaptoethanol, thioglycerol and α -mercapto- ω -hydroxy-oligoethylene oxides are particularly preferred.

25

30

Preferred hydroxyl group-containing mercaptocarboxylic esters are esters of mercaptoacetic acid, mercaptopropionic acid and mercaptobutyric acid with ethylene

09980615.011802

glycol, propylene glycol, butylene glycol, diethylene glycol, triethylene glycol, tetraethylene glycol, octaethylene glycol, dipropylene glycol, tripropylene glycol, tetrapropylene glycol, glycerol, N-methyl-diethanolamine. The corresponding esters of mercaptoaetic acid and 3-mercaptopropionic acid are particularly preferred.

5

Suitable radical starters for the addition of hydroxylmercaptans to hydroxyl group-containing rubbers are e.g. azo-initiators such as azobisisobutyronitrile, azobiscyclohexanonitrile and peroxides such as dilauroyl peroxide, benzopinacolsilyl ether or photoinitiators in the presence of UV or visible light. Preferred radical starters are diacyl peroxides, in particular dilauroyl peroxide, didoceanoyl peroxide, di-(3,3,5-trimethylhexanoyl) peroxide, disuccinoyl peroxide, dibenzoyl peroxide and perketals such as 1,1-di-(tert.-butylperoxy)-3,3,5-trimethylcyclohexane, 1,1-di-(tert.-butylperoxy)-cyclohexane and 1,1-di-(tert.-butylperoxy)-butane.

10

15 Preferred amounts of radical starters are 0.5 to 20 wt.%, with respect to the hydroxylmercaptan.

20

Sub AG → The average molecular weight of the hydroxyl group-containing rubbers is between 50 000 and 2 000 000, preferably between 100 000 and 1 000 000.

25

The Mooney viscosity ML 1+4 of the copolymers is between 10 and 200, preferably 30 and 150, measured at 100°C.

30

The concentration of copolymerised 1,2-butadiene units ("vinyl content") is between 0 and 60 wt.%, preferably 1 to 50 wt.%.

The glass transition temperature is between -120 and -50°C, preferably -120 and -70°C. The glass transition temperature can be determined using known methods, e.g. by means of DSC (differential scanning calorimetry, rate of heating 20 K/min).

30

09980615-011802

The cis-1,4 content of copolymerised dienes is between 10 and 100 %, preferably between 30 and 99 %, in particular 90 - 99 %.

5 The concentration of primary hydroxyl groups is between 0.1 and 2 wt.%, preferably in the range 0.1 to 1 wt.%, in particular in the range 0.1 to 0.75 wt.%, with respect to the rubber.

10 The concentration of hydroxyl groups can be determined using known methods, that is, for example, by spectroscopy, titrimetry, elemental analysis or by determining the so-called hydroxyl value (OH value), that is by reaction with reagents which release titratable acids on contact with OH groups. See DIN 53 240 for details.

15 Rubbers according to the invention with a glass transition temperature of -120 to -50°C and 0.1 to 2 wt.% of hydroxyl groups may be used individually, blended with aromatic or aliphatic oils or mixed with other rubbers. Additional rubbers which are suitable for the production of rubber vulcanisates are, in addition to natural rubber, synthetic rubbers. Preferred synthetic rubbers are described, for example, in W. Hoffmann, Kautschuktechnologie, Gentner Verlag, Stuttgart 1980 and I. Franta, Elastomers and Rubber Compounding Materials, Elsevier, Amsterdam, 1989. They
20 include, inter alia,

- BR - polybutadiene
- ABR - butadiene/C₁-C₄-alkyl acrylate copolymers
- CR - polychloroprene
- 25 IR - polyisoprene
- SBR - styrene/butadiene copolymers with styrene contents of 1 to 60, preferably 20 to 50 wt.%

09980615-011802

- IIR - isobutylene/isoprene copolymers
- NBR - butadiene/acrylonitrile copolymers with acrylonitrile contents
of 5 to 60, preferably 10 to 40 wt. %
- HNBR - partially hydrogenated or fully hydrogenated NBR rubber
- 5 EPDM - ethylene/propylene/diene copolymers

and mixtures of these rubbers. For producing vehicle tires, natural rubber, emulsion
SBR and solution SBR rubbers with a glass transition temperature above -50°C ,
polybutadiene rubber with a high 1,4-cis content ($>90\%$), which has been prepared
10 using catalysts based on Ni, Co, Ti or Nd, and polybutadiene rubber with a vinyl
content of 0 to 75 % and their mixtures are of particular interest.

Quite specifically preferred rubber mixtures according to the invention contain, in
addition to the hydroxyl group-containing rubber with a glass transition temperature
15 between -120° and -50°C . additional rubbers from the group comprising natural rub-
ber, polyisoprene and styrene/butadiene copolymers with styrene contents between
10 and 50 wt. %. The amount of these additional rubbers is normally in the range 0.5
to 95, preferably 40 to 90 wt. %, with respect to the entire amount of rubber in the
rubber mixture. The amount of additionally added rubbers is again governed by the
20 particular ultimate use of the rubber mixtures according to the invention.

Rubber mixtures according to the invention contain 5 to 300 parts by wt. of an active
or inactive filler such as, for example:

- 25 - highly disperse silicas prepared e.g. by precipitation from solutions of sili-
cates or by flame hydrolysis of silicon halides with specific surface areas of 5
to 1000, preferably 20 to 400, m^2/g (BET surface area) and with primary par-
ticle sizes of 10 to 400 nm. The silicas may also optionally be present as
mixed oxides with other metal oxides such as Al, Mg, Ca, Ba, Zn, Zr or Ti
30 oxides,

- synthetic silicates such as aluminium silicate or alkaline earth metal silicates such as magnesium silicate or calcium silicate, with BET surface areas of 20 to 400 m²/g and primary particle diameters of 10 to 400 nm,
- 5 - natural silicates such as kaolin and other naturally occurring silicas,
- glass fibre and glass fibre products (mats, ropes) or glass microbeads,
- metal oxides such as zinc oxide, calcium oxide, magnesium oxide, aluminium oxide,
- 10 - metal carbonates such as magnesium carbonate, calcium carbonate, zinc carbonate,
- 15 - metal hydroxides such as e.g. aluminium hydroxide, magnesium hydroxide,
- carbon blacks. The carbon blacks to be used here are prepared by the lamp black, furnace black or channel black processes and have BET surface areas of
- 20 20 to 200 m²/g, such as e.g. SAF, ISAF, HAF, FEF or CPF carbon blacks,
- rubber gels, in particular those based on polybutadiene, butadiene/styrene copolymers, butadiene/acrylonitrile copolymers and polychloroprene.
- 25 Highly disperse silicas and carbon blacks are preferably used as fillers.

The fillers mentioned may be used individually or as a mixture. In a particularly preferred embodiment, the rubber mixtures contain a mixture of pale-coloured fillers, such as highly disperse silicas, and carbon blacks, as filler, wherein the mixing ratio

30 of pale-coloured fillers to carbon blacks is 0.05 to 20, preferably 0.1 to 10.

09980615-011802

The fillers are preferably added to the solution of polymerised rubber as solids or as a slurry in water or in a solvent to dissolve the hydroxyl group-containing rubbers(s). The rubber solution may be prepared beforehand, but the solution arising from polymerisation is preferably used directly. The solvent is then removed thermally or, preferably with the assistance of steam. The conditions for the stripping process can readily be determined in preliminary trials.

Furthermore, the fillers are preferably added to solid hydroxyl group-containing rubber or a mixture of rubbers and incorporated therein in a known manner, e.g. with a compounder.

Furthermore, rubber mixtures according to the invention also contain optional cross-linking agents. Sulfur or peroxides may be used as cross-linking agents, wherein sulfur is particularly preferred. Rubber mixtures according to the invention may contain further rubber auxiliary substances such as reaction accelerators, antioxidants, thermal stabilisers, light stabilisers, anti-ozonants, processing aids, plasticisers, tackifiers, blowing agents, colorants, pigments, waxes, extenders, organic acids, delaying agents, metal oxides and activators such as triethanolamine, polyethylene glycol, hexanetriol, etc. which are known in the rubber industry.

In preferred rubber mixtures with highly active precipitated silicas, the use of additional filler activators is particularly advantageous. Preferred filler activators are sulfur-containing silyl ethers, in particular bis-(trialkyloxysilyl-alkyl)-polysulfides, as are described in DE 2,141,159 and DE-AS 2,255,577, the oligomers and/or polymers of sulfur-containing silyl ethers in DE-OS 4,435,311 and EP-A 670,347, mercaptoalkyltrialkoxysilanes, in particular mercaptopropyltriethoxysilane and thiocyanatoalklyl silyl ethers such as are described in DE-OS 19,544,469.

The rubber auxiliary substances are used in conventional amounts, governed, inter alia, by the ultimate use. Conventional amounts are e.g. amounts of 0.1 to 50 wt.%, with respect to rubber.

09980615-011377

Rubber mixtures according to the invention are outstandingly suitable for producing moulded items of all types.

- 5 Non-restricting examples of these moulded items are O-rings, sections, seals, membranes, tires, tire treads, damping elements and hoses.

Tires and tire treads are particularly preferred.

09980615.011P02

Examples

Example 1

5 12.5 g of 1-mercapto-2-ethanol and 1 g of dilauroyl peroxide were added to a solution of 500 g of solution polybutadiene rubber, Buna CB 65 (Bayer AG, Li type, cis-1,4 content about 40 %) in 4 l of cyclohexane at 70°C. The mixture was then stirred for 8 hours at 80°C. At this point about 39 % of the mercaptoethanol had reacted. Then 2.5 g of antioxidant Vulkanox 4020 (Bayer AG) were added and the solvent
10 was distilled off using steam (100-110°C). After drying at 70°C under vacuum, 508 g of a colourless rubber with an OH-value of 7, an OH content of 0.21 wt.% and a cis-1,4 content of 40 % were obtained. The glass transition temperature was -90°C.

Example 2

15 The same method was used as described in example 1, but only 6.25 g of 1-mercapto-2-ethanol were used. After 8 hours at 80°C, 41 % conversion was achieved. Then the solvent was distilled off using steam (100-110°C). After drying at 70°C under vacuum, 505 g of a colourless rubber with a OH content of 0.1 wt.% and a cis-
20 1,4 content of 40 % were obtained. The glass transition temperature was -90°C.

Example 3

25 12.5 g of 1-mercapto-2-ethanol and 1 g of dilauroyl peroxide were added to a solution of 500 g of solution polybutadiene rubber Buna VI 47-0 (Bayer AG, vinyl-BR, concentration of 1,2-bonded butadiene (vinyl content) about 47 %) in 4 l of cyclohexane at 70°C. The mixture was then stirred for 4 hours at 80°C. At this point, 95 % of the mercaptoethanol had reacted. Then 2.5 g of antioxidant Vulkanox 4020 (Bayer AG) were then added and the solvent was distilled off using steam (100-110°C).
30 After drying under vacuum at 70°C, 512 g of a colourless rubber with an OH value of

17, an OH content of 0.5 wt.%, a vinyl content of about 45 % and a glass temperature of -54°C were obtained.

09980615-011802